Identification of an unknown biocide by volatile atmospheric pressure chemical ionization (vAPCI) and compact mass spectrometry (CMS)



Overview:

A rapid unknown sample analysis is performed using a novel volatile APCI source with CMS detection and on-line database search.

Introduction:

Volatile sampling and APCI generation of ions (vAPCI) is characterized by a spatial separation between the APCI sample ionization in the MS source region and the analyte sampling through a flexible and heated transfer line (Figure 1). vAPCI allows headspace sampling over targets in various aggregate states such as gaseous solids or liquids. samples, vAPC been predominantly used for highly volatile analytes such as flavor components of food stuffs (spoiled meats or cheeses) analysis^[1]; breath and however, vAPCI can also be used for other applications in a rapid sample screening for detection fashion ot unknowns in 'white powders'.

Methods and Materials:

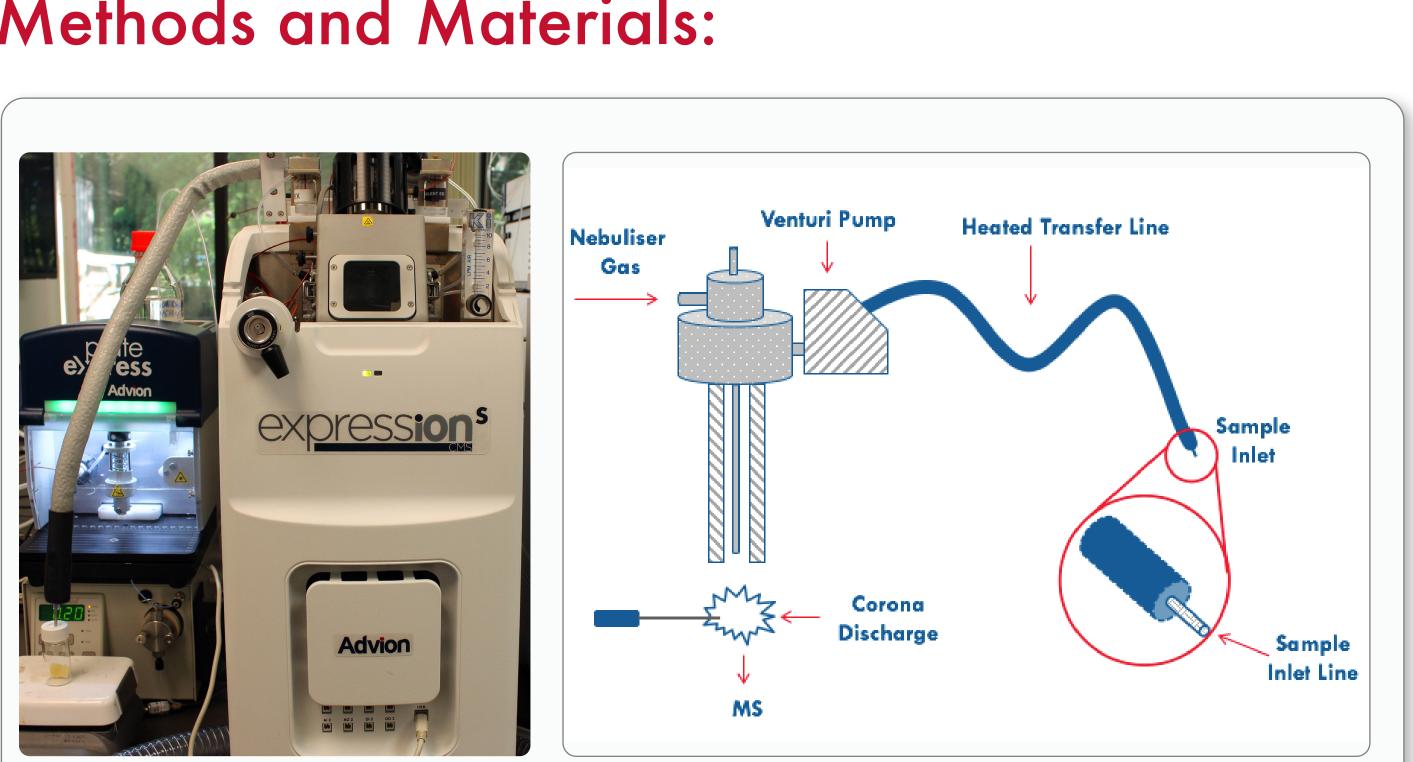


Figure 1: A vAPCI system set up with corresponding schematic. The system is coupled with the APCI source of the expression^s CMS. This includes the 120° C heated transfer line in which the headspace is moved through the system via the venturi effect.

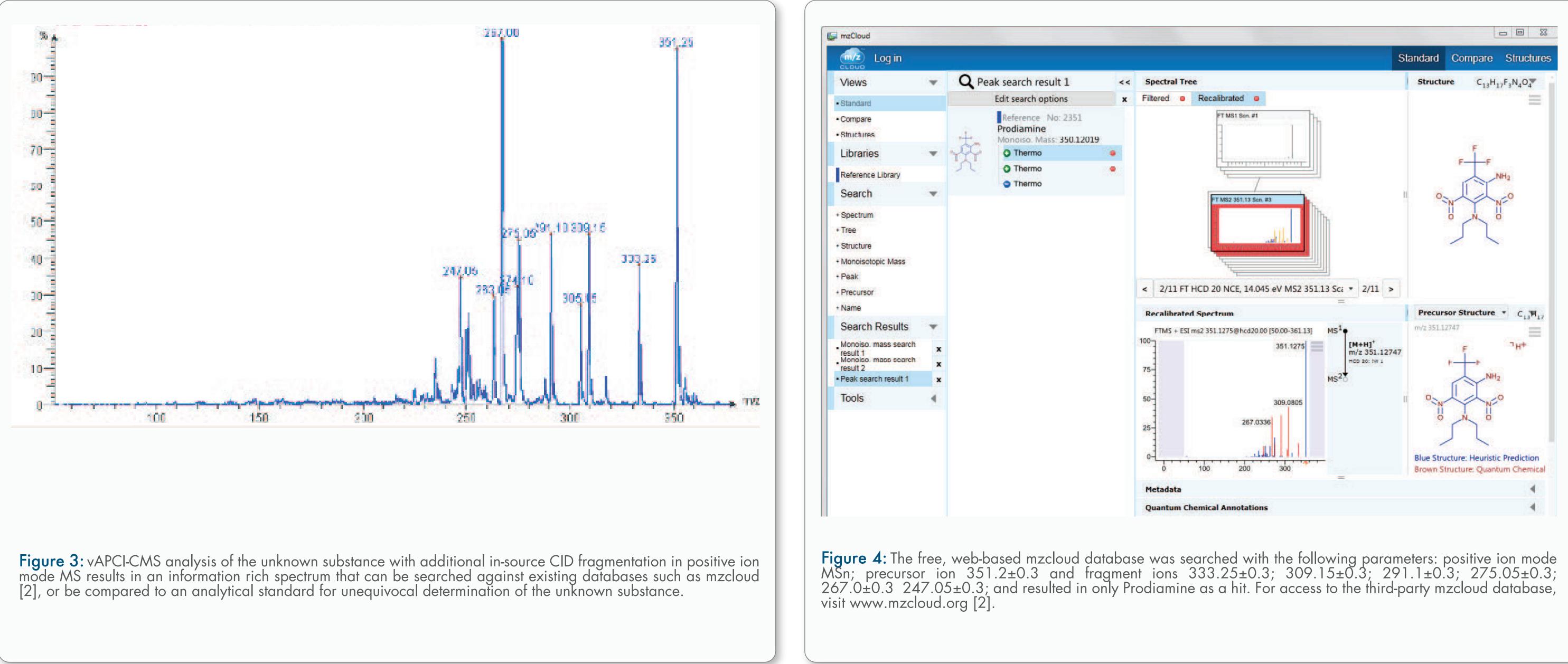
50 mg unknown lawn treatment pellets (Figure collected, were crushed and heated to ca. 50°C. The headspace transported Was venturi effect through a 120°C heated transfer line towards the APCI source of the expression^S



compact mass spectrometer (Advion Inc.). Two different ion source settings were utilized to collect data for positive and negative ion mode as well as in-source fragmentation for both. The data was analyzed for mass-to-charge ratio, isotopic pattern and fragmentation information and searched against the free online mzcloud database (Highchem LLC - https://www.mzcloud.org/)^[2].

Results:

- over that could be avoided at 50°C.
- mono-isotopic mass could therefore be assumed to be Mw(iso) 350.15.
- 291.10, 275.05, 267.05 and 247.05 (Figure 3).



• Crushing and heating the sample to ca. 50°C significantly increased signal intensity, however temperatures above 60°C resulted in carry

MS analysis showed a strong signal at m/z 351.15 in positive ion mode and m/z 349.15 in negative ion mode, both signals showed an isotope distribution indicating a singly charged molecule and not indicating any halogen atoms present. The unknown substance

Further in-source fragmentation showed an informative spectrum only in positive ion mode with strong signals at m/z 333.25, 309.15;

Two databases were queried with this information; the • off-line excel based herbicide specific database^[3] resulted in two potential herbicides; Prodiamine and Cafenstrole as likely candidates for the unknown sample (data not shown). The only hit returned from the mzcloud database^[2] was Prodiamine with an excellent correlation and additional overlap of fragment ions (Figure 4).

In combination with additional background information • (Prodiamine being a NY state approved pre-emergent lawn herbicide) and comparison to an analytical (data not shown); Prodiamine can standard unequivocally be determined as the active ingredient of the previously unknown sample.

Conclusions:

vAPCI allows measurement from the headspace of • unknown samples in gas, liquid or solid state (eg white powder) with little to no sample preparation or carry over.

CMS detection combined with polarity switching, • in-source fragmentation and database searches is a versatile analytical tool to identify unknown compounds

Literature and Acknowledgements:

¹ Heaney LM, Ruszkiewicz DM, Arthur KL, Hadjithekli A, Aldcroft C, Lindley MR, Thomas CLP, Turner MA and Reynolds JC: Real-time monitoring of exhaled volatiles using atmospheric pressure chemical ionization on a compact mass spectrometer. Bioanalysis 2016 doi/full/10.4155/bio-2016-0045

^[2] ESI-MS/MS spectrum of Prodiamine from m/z cloud database for comparison: https://www.mzcloud.org/DataViewer#Creference2351#T3285#c#318704

^[3] Gandy M, Corral M, Mylne J and Stubbs K: An interactive database to explore herbicide physicochemical properties. Organic and 5586-5590; Biomolecular 2015 13(20) Chemistry http://www.mylne.org/herbicides.html

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